

2-DIPHENYLMETHYLSILYLETHYL GROUP AS A NEW PROTECTING GROUP OF INTERNUCLEOTIDIC PHOSPHATES IN OLIGONUCLEOTIDE SYNTHESIS

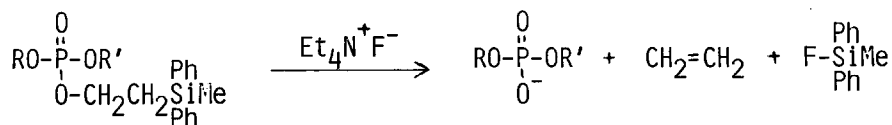
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**Abstract:** Internucleotidic phosphates were protected by 2-diphenylmethylsilyl-ethyl group which was selectively removed by treatment with tetrabutylammonium fluoride.

There have been several successful reports on the synthesis of oligonucleotides via phosphotriester approach,<sup>1)</sup> however, there still remain crucial problems of the removal of protecting groups of internucleotidic phosphates.<sup>2)</sup> The problem may be solved to find more suitable protecting group which can be removed by  $\beta$ -elimination mechanism from the internucleotidic phosphates. Up to date, there have been a few phosphate protecting groups which satisfy the above requirement. For example, 2-cyanoethyl,<sup>3)</sup> 2,2,2-trichloroethyl,<sup>4)</sup> and 2-p-nitrophenylethyl<sup>5)</sup> groups were proposed and examined in several laboratories.

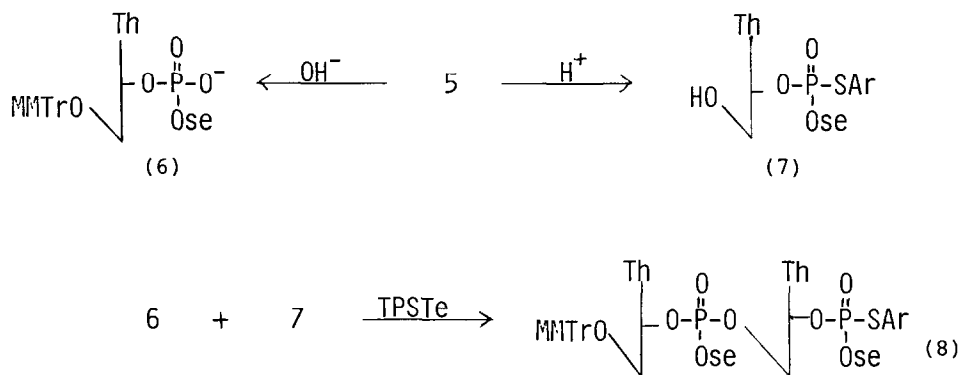
On the other hand, Gerlach<sup>6)</sup> and Carpino<sup>7)</sup> reported that 2-trimethylsilyl-ethyl group could be used for protection of carboxylic acids and the ester linkage could be easily cleaved by treatment with tetrabutylammonium fluoride (TBAF). 2-Trimethylsilylpropen-2-yl group was proposed by Chan<sup>8)</sup> and applied to simple phosphates. We have tried to apply these protecting groups containing silicon atom to the synthesis of oligonucleotides. However, they were found to be too labile under any conditions for the condensation reactions. In order to overcome the instability of such a type of protecting groups containing silicon atom, several 2-tri-alkyl(or aryl)silylethyl groups were examined. Finally, we could find 2-diphenylmethylsilylethyl group (se) which could be introduced easily and removed selectively by treatment with TBAF from the phosphotriesters along with the formation of ethylene and di-phenylmethylsilyl fluoride.



A new phosphorylating agent, 2-diphenylmethylsilylethyl S-p-methoxyphenyl

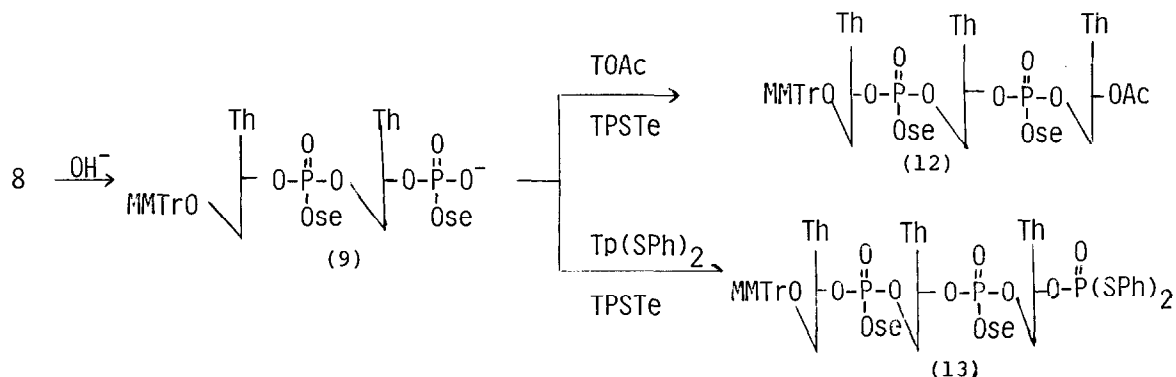


protected dithymidylate, MMTrTp(se)Tp(se)(SAr), (8) in 72% (108 mg) yield (Rf. 0.42, solvent A). In the above reaction other condensing agents, e.g., TPS and mesitylenesulfonyltriazole were found to be ineffective.



Similarly, the fully protected dinucleoside monophosphates, MMTrTp(se)TOAc (10) and MMTrA<sup>bz</sup>p(se)TOAc (11) were synthesized in 60% and 72% yield, respectively. In both cases, the diastereoisomers were separated on tlc (see Table 1).

Further, the fully protected trinucleotides e.g., MMTrTp(se)Tp(se)TOAc (12) and MMTrTp(se)Tp(se)Tp(SPh)<sub>2</sub> (13) were synthesized in 55% and 47% yields respectively. The reaction conditions and the results are summarized in Table 1.



Deprotection of 11 was performed as follows: Compound 11 was treated with 0.05 M TBAF (2 equiv.) in (THF/pyridine/water-8:1:1) at room temperature for 24 h, followed by treatment of methanolic ammonia and 80% acetic acid in the usual workup. Dinucleoside phosphate, ApT, was obtained by paper chromatography (Whatman 3MM) in almost quantitative yield. The structure of ApT was confirmed by the enzymatic degradation with snake venom phosphodiesterase, giving a reasonable ratio of Ap:T (1.13:1:0).

Deprotection of both 10 and 12 was performed in the same manner, to afford TpT and TpTpT in almost quantitatively.

Table 1 Synthesis of the Fully Protected Di- and Tri-nucleotides

Phosphate component (mmol)	hydroxyl component (mmol)	condensing agent (mmol)	pyridine (ml)	time (h)	product (yield)	Rf value (solvent A*)
MMTrTp(se) (0.26)	TOAc (0.20)	MSTe*** (0.82)	1.5	3.5	10 (72%)	0.68 and 0.58**
MMTrA <sup>bz</sup> <sub>p</sub> (se) (0.41)	TOAc (0.33)	TPSTe (1.22)	2.0	5	11 (60%)	0.49 and 0.40**
MMTrTp(se)Tp(se) (0.07)	TOAc (0.09)	TPSTe (0.21+0.14)	1.0	40	12 (55%)	0.30
MMTrTp(se)Tp(se) (0.085)	Tp(SPh) <sub>2</sub> (0.08)	TPSTe (0.424)	2.0	37	13 (47%)	0.37

\* Solvent A : CH<sub>2</sub>Cl<sub>2</sub>/MeOH, 9:1 (v/v)

\*\* diastereoisomers; \*\*\* MSTe refers to mesitylenesulfonyltetrazole.

In these cases, the cleavage of internucleotide bond didn't take place during the deprotection since any other spots were not detected on paper chromatogram.

Compared with other protecting groups, it is noted that the se group can be easily removed by an attack of fluoride ion from TBAF not on phosphorus atom but on silicon atom via  $\beta$ -elimination mechanism, so that reactive intermediate, the phosphorofluoridate does not produced during the deprotection process.

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